# Giant magnetocaloric effect and magneto-crystalline coupling

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# Outlook

Magneto-crystalline coupling

How to change the coupling: Fe<sub>2</sub>P chemical pressure x physical pressure Fe<sub>2</sub>P to (Fe,Mn)<sub>2</sub>(P, ?) similarities and differences What information can we obtain from it?

Conclusions



### **Magneto-crystalline coupling**

magnetic interaction is

dependent on interatomic distance



![](_page_3_Picture_0.jpeg)

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# Chemical pressure: Fe<sub>2</sub>P

![](_page_3_Figure_2.jpeg)

![](_page_4_Picture_0.jpeg)

# Chemical pressure: Fe<sub>2</sub>P

![](_page_4_Figure_3.jpeg)

![](_page_5_Picture_0.jpeg)

![](_page_5_Figure_2.jpeg)

y

1/4

1/4

![](_page_5_Figure_3.jpeg)

Fe - tetrahedral 0.5  $\mu_B$ Mn - pyramidal 2.6  $\mu_B$ 

Watanabe et al. JPSJ 1973

	Interatomic distance Å	Sign of exchange interaction		
Fe-Fe	2.631	antiferromagnetic		
Fe-Mn	2.661 2.766 2.79	critical* (ferromagnetic) antiferromagnetic critical*		
Mn-Mn	3.05	antiferromagnetic		

\* "critical" means that the interaction is in the region where the sign changes from antiferromagnetic to ferromagnetic.

antiferromagnetic Co<sub>2</sub>P - orthorhombic

![](_page_6_Picture_0.jpeg)

### Chemical pressure $FeMnP_{1-x}A_x$ A = As, Ge, Si

Nominal, x	0.25	0.35	0.45	0.50	0.53	0.55(I)	0.65
a (Å)	6.0392	6.0677	6.1080	6.1290	6.1628	6.1739	6.2120
<i>c</i> (Å)	3.4870	3.4874	3.4900	3.4805	3.4946	3.4511	3.4633
c/a	0.5774	0.5748	0.5714	0.5679	0.5670	0.5590	0.5575
$T_{\rm C}({\rm K})$	168	213	240	282	290	300(I)	332
						307(II)	

![](_page_6_Figure_4.jpeg)

MnFeP<sub>1-x</sub>As<sub>x</sub>

ferromagnetism and Fe<sub>2</sub>P structure are recovered

Tegus, PhD thesis University of Amsterdam

![](_page_7_Picture_0.jpeg)

### Chemical pressure FeMnP<sub>1-x</sub>A<sub>x</sub> A = As, Ge, Si

У	c/a	$\Delta T_{ m hys}$ (K)	Т <sub>с</sub> (К)	$-\Delta S_{m,\max}$ (J kg <sup>-1</sup> K <sup>-1</sup> )	RCP (J kg <sup>-1</sup> )
0.80	0.5626	1	288	20.3	151
0.78	0.5638	2	274	15.3	162
0.76	0.5646	2	254	16.4	151
0.70	0.5651	0	230	9.8	155
Gadolini	um <sup>a</sup>	0	293	4.2	166

![](_page_7_Figure_3.jpeg)

*Mn<sub>2-y</sub>Fe<sub>y</sub>P<sub>0.75</sub>Ge<sub>0.25</sub>* 

# ferromagnetism and Fe<sub>2</sub>P structure are recovered

Appl. Phys. Lett. 94, 102513 (2009)

![](_page_8_Picture_0.jpeg)

![](_page_8_Figure_3.jpeg)

![](_page_9_Picture_0.jpeg)

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# Probing the coupling with pressure

![](_page_9_Figure_2.jpeg)

Fe<sub>2</sub>P- like behavior

![](_page_10_Picture_0.jpeg)

![](_page_10_Figure_3.jpeg)

![](_page_11_Picture_0.jpeg)

![](_page_11_Figure_2.jpeg)

![](_page_12_Picture_0.jpeg)

# Probing the coupling with pressure

 $Mn_{1.16}Fe_{0.84}P_{0.75}Ge_{0.25}$ 

![](_page_12_Figure_3.jpeg)

![](_page_13_Picture_0.jpeg)

![](_page_13_Figure_3.jpeg)

![](_page_14_Picture_0.jpeg)

### Why no change in $T_C$ ?

![](_page_14_Figure_3.jpeg)

![](_page_15_Picture_0.jpeg)

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### Why no change in $T_C$ ?

### $Fe_{0.84}Mn_{1.16}P_{0.75}Ge_{0.25}$

![](_page_15_Figure_3.jpeg)

![](_page_16_Picture_0.jpeg)

### **Conclusions?**

- Fe<sub>2</sub>P-based compounds present different behaviors under pressure
- chemical pressure is not always equivalent to physical pressure
- how to understand the entire picture?

![](_page_17_Picture_0.jpeg)

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### Thank you for your attention!