Determining the magnetocaloric effect in hysteretic materials

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The magnetocaloric effect



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Measurement

Direct measurements - ΔS and ΔT_{ad}

Indirect measurements – calculate ΔS from isothermal magnetization

Maxwell relations

$$\left(\frac{\partial S(T,H)}{\partial H}\right)_{T} = \left(\frac{\partial M(T,H)}{\partial T}\right)_{H} \rightarrow \Delta S_{T}(T)_{\Delta H} = \int_{H_{1}}^{H_{2}} \left(\frac{dM}{dT}\right)_{H} dH$$

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Reasons for questioning the validity of the Maxwell relations



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Colossal Magnetocaloric Effect



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Validity of the Maxwell relations

2nd order – Maxwell is valid 1st order – Clausius-Clapeyron or Maxwell?

$$\frac{dH}{dT} = -\frac{\Delta S_M}{\Delta M} \Leftrightarrow \left(\frac{\partial S}{\partial H}\right)_{T,P} = \left(\frac{\partial M}{\partial T}\right)_{H,P}$$

Sun et al. PRL 85 4191

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Liu's solution



Liu et al. Appl. Phys. Lett. 90 032507

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What are we measuring?



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$\partial S(T,H)$) _	$\left(\frac{\partial I}{\partial I}\right)$	M(T, H)	H)
∂H	\int_T	-	∂T	$ \int_{H} $



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Conclusions

The entropy change calculated from magnetic isothermal measurements using the Maxwell relation is a very good tool around 1st order magnetic phase transitions as long as the history of the sample is taken into account when planning a measurement.

