

## **APPENDIX for the Examination Co-ordinating Radiation Protection Expert**

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Examination date: 12 December 2016  
Duration of examination: 13:30 - 16:30

<b>Instructions:</b>
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- ❑ **If you use any data other than the data provided in this appendix, please state the source!**
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*Handboek Radionucliden*, [Radionuclides Handbook], A.S. Keverling Buisman (2nd edition 2007), pp. 26-27,  $^{18}\text{F}$  data.

$^{18}\text{F}$

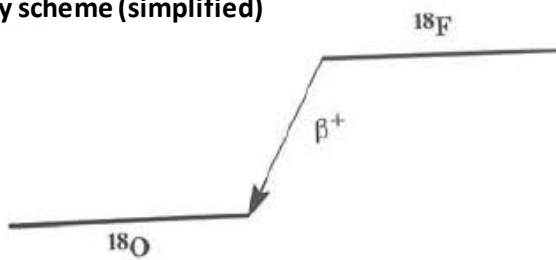
$Z = 9$

**Half-life and decay constant**

$T_{1/2} = 109,70 \text{ min} = 6,582 \times 10^3 \text{ s}$

$\lambda = 1,05 \times 10^{-4} \text{ s}^{-1}$

**Decay scheme (simplified)**



**Main emitted radiation**

Straling	$y \text{ (Bq}\cdot\text{s)}^{-1}$	$E \text{ (keV)}$
$\beta^+$	1,000	250   634
$\gamma^\pm$	2,000	511

**Source constants**

Air kerma rate  
Ambient dose equivalent rate

$k = 0,135 \text{ }\mu\text{Gy/h per MBq/m}^2$   
 $h = 0,166 \text{ }\mu\text{Sv/h per MBq/m}^2$

**Miscellaneous**

Specific activity  
Exemption levels  
Skin contamination  
Wound contamination/injection  
Transport

$A_{sp} = 3,52 \times 10^{18} \text{ Bq/g}$   
 $C_v = 10^1 \text{ Bq/g en } A_v = 10^6 \text{ Bq}$   
 $H_{\text{huid}} = 5 \times 10^{-10} \text{ Sv/s per Bq/cm}^2$   
 $e(50) = 1,4 \times 10^{-11} \text{ Sv/Bq}$   
 $A_1 = 1 \text{ TBq}$   
 $A_2 = 0,6 \text{ TBq}$

**Production and applications**

Het radionuclide  $^{18}\text{F}$  is een cyclotronproduct. Het nuclide wordt toegepast in de nucleaire geneeskunde voor het maken van afbeeldingen met behulp van positronen-emissie-tomografie (PET).

N = 9

 $^{18}\text{F}$ **Metabolic model**

Voor stralingshygiënische doeleinden wordt aangenomen dat fluor na opname in het bloed volledig in het bot wordt opgenomen.  
Gezien de korte fysische halveringstijd van  $^{18}\text{F}$  (110 minuten) is de biologische halveringstijd van geen belang.

**Ingestion and lung clearance classes**

Ingestie			
Alle verbindingen	$f_1 = 1$		
Inhalatie			
Afhankelijk	$f_1 = 1$		Klasse S
van	$f_1 = 1$		Klasse M
bindingskation	$f_1 = 1$		Klasse F

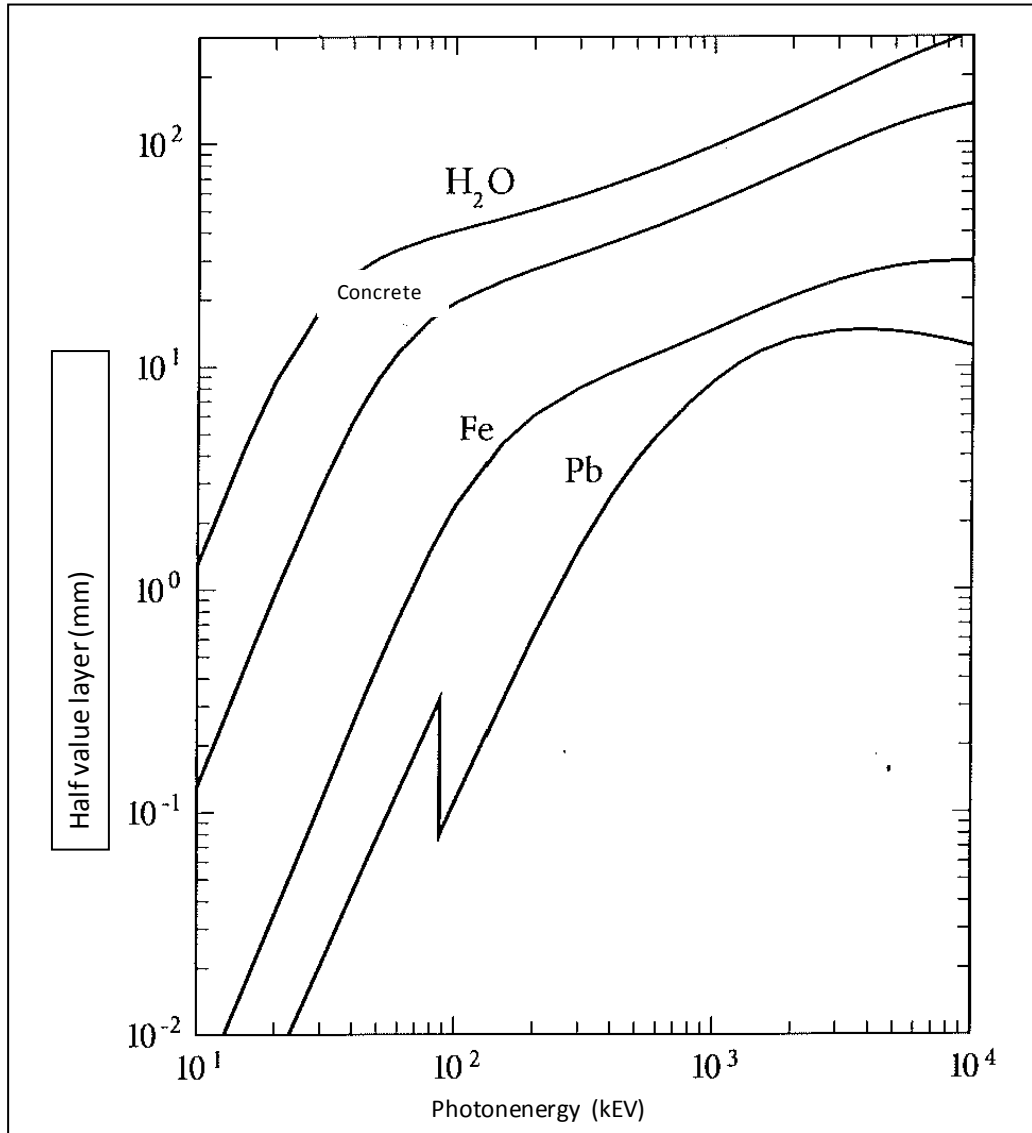
**Dose conversion coefficient and radiotoxicity equivalent for workers (w) and members of the public (b)**

	Ingestie $f_1 = 1$	Inhalatie F	Inhalatie M	Inhalatie S	
$e(50)(w)$	$4,9 \times 10^{-11}$	$5,4 \times 10^{-11}$	$8,9 \times 10^{-11}$	$9,3 \times 10^{-11}$	Sv/Bq
$A_{Re}(w)$	$2,0 \times 10^{10}$	$1,9 \times 10^{10}$	$1,1 \times 10^{10}$	$1,1 \times 10^{10}$	Bq
$e(50)(b)$	$4,9 \times 10^{-11}$	$3,0 \times 10^{-11}$	$5,7 \times 10^{-11}$	$6,0 \times 10^{-11}$	Sv/Bq
$A_{Re}(b)$	$2,0 \times 10^{10}$	$3,3 \times 10^{10}$	$1,8 \times 10^{10}$	$1,7 \times 10^{10}$	Bq

**Data for total body counting (after single intake)**

Time (d)	Total body activity (Bq per Bq intake)			
Tijd (d)	Lichaamsactiviteit (Bq per Bq inname)			
0,25	$1,0 \times 10^{-1}$	$7,7 \times 10^{-2}$	$7,7 \times 10^{-2}$	$7,7 \times 10^{-2}$
1	$1,1 \times 10^{-4}$	$6,8 \times 10^{-5}$	$6,8 \times 10^{-5}$	$6,8 \times 10^{-5}$

**Figure 2: Half-value layer of different shielding materials for narrow-beam photon radiation**



**Table 1: Broadbeam transmission factors at 511 keV (AAPM Task Group 108)**

Thickness <sup>a, b</sup>	Transmission Factors		
	Lead	Concrete <sup>c</sup>	Iron
0	1.0000	1.0000	1.0000
1	0.8912	0.9583	0.7484
2	0.7873	0.9088	0.5325
3	0.6905	0.8519	0.3614
4	0.6021	0.7889	0.2353
5	0.5227	0.7218	0.1479
6	0.4522	0.6528	0.0905
7	0.3903	0.5842	0.0542
8	0.3362	0.5180	0.0319
9	0.2892	0.4558	0.0186
10	0.2485	0.3987	0.0107
12	0.1831	0.3008	0.0035
14	0.1347	0.2243	0.0011
16	0.0990	0.1662	0.0004
18	0.0728	0.1227	0.0001
20	0.0535	0.0904	
25	0.0247	0.0419	
30	0.0114	0.0194	
40	0.0024	0.0042	
50	0.0005	0.0009	

Broadbeam transmission factors at 511 keV in lead, concrete, iron.

<sup>a</sup>Thickness in mm for lead.

<sup>b</sup>Thickness in cm for concrete and iron.

<sup>c</sup>Concrete density = 2.35 g / cm<sup>3</sup> .

**Table 2: Build-up factors for an isotropic point source**

materiaal	Fotonenenergie E (MeV)	$\mu d$						
		1	2	4	7	10	15	20
water	0,255	3,09	7,14	23,0	72,9	166	456	982
	0,5	2,52	5,14	14,3	38,8	77,6	178	334
	1,0	2,13	3,71	7,68	16,2	27,1	50,4	82,2
	2,0	1,83	2,77	4,88	8,46	12,4	19,5	27,7
	3,0	1,69	2,42	3,91	6,23	8,63	12,8	17,0
aluminium	0,5	2,37	4,24	9,47	21,5	38,9	80,8	141
	1,0	2,02	3,31	6,57	13,1	21,2	37,9	58,5
	2,0	1,75	2,61	4,62	8,05	11,9	18,7	26,3
	3,0	1,64	2,32	3,78	6,14	8,65	13,0	17,7
ijzer	0,5	1,98	3,09	5,98	11,7	19,2	35,4	55,6
	1,0	1,87	2,89	5,39	10,2	16,2	28,3	42,7
	2,0	1,76	2,43	4,13	7,25	10,9	17,6	25,1
	3,0	1,55	2,15	3,51	5,85	8,51	13,5	19,1
lood	0,5	1,24	1,42	1,69	2,00	2,27	2,65	2,73
	1,0	1,37	1,69	2,26	3,02	3,74	4,81	5,86
	2,0	1,39	1,76	2,51	3,66	4,84	6,87	9,00
	3,0	1,34	1,68	2,43	3,75	5,30	8,44	12,3

**Handboek Radionucliden [Radionuclides Handbook], A.S. Keverling Buisman (2nd edition, 2007), pp. 42-43, <sup>40</sup>K data**

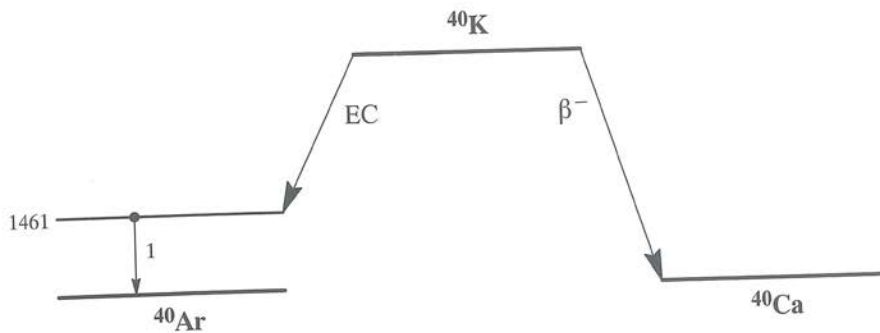
40K Z = 19

**Half-life and decay constant**

$T_{1/2} = 1,277 \times 10^9 \text{ j} = 4,03 \times 10^{16} \text{ s}$

$\lambda = 1,72 \times 10^{-17} \text{ s}^{-1}$

**Decay scheme (simplified)**



**Main emitted radiation**

Straling	y (Bq·s) <sup>-1</sup>	E (keV)
β <sup>-</sup>	0,893	585   1312
γ <sub>1</sub>	0,107	1461
KLL	0,065	3

**Source constants**

Air kerma rate	$k = 0,018 \text{ } \mu\text{Gy/h per MBq/m}^2$
Ambient dose equivalent rate	$h = 0,021 \text{ } \mu\text{Sv/h per MBq/m}^2$

**Miscellaneous**

Specific activity	$A_{sp} = 2,59 \times 10^5 \text{ Bq/g}$
Exemption levels	$C_v = 10^2 \text{ Bq/g}$ en $A_v = 10^6 \text{ Bq}$
Skin contamination	$H_{huid} = 5 \times 10^{-10} \text{ Sv/s per Bq/cm}^2$
Wound contamination / injection	$e(50) = 5,1 \times 10^{-9} \text{ Sv/Bq}$
Transport	$A_1 = 0,9 \text{ TBq}$ $A_2 = 0,9 \text{ TBq}$

**Production and applications**

Kalium-40 is een primordiaal radionuclide, dat wil zeggen het is sinds de nucleogenese (het moment van ontstaan van de zwaardere nucliden) nog niet vervallen. Het komt voor



N = 21

<sup>40</sup>K

in het menselijk lichaam, via ingestie. De <sup>40</sup>K-activiteit in de referentiemens bedraagt 4000 Bq. Natuurlijk kalium heeft een <sup>40</sup>K-gehalte van 0,0118% (at), hetgeen leidt tot een specifieke activiteit van kalium van 30 Bq/g. Het nuclide wordt veel gebruikt als ijkbron, aangezien de activiteit door weging kan worden gevonden.

### Metabolic model

Voor stralingshygiënische doeleinden wordt aangenomen dat kalium zich vanuit het bloed homogeen over alle organen/weefsels verdeelt.

De biologische halveringstijd voor deze organen/weefsels wordt gesteld op 30 dagen.

### Ingestion and lung clearance classes

Ingestie			
Alle verbindingen	$f_1 = 1$		
Inhalatie			
Alle verbindingen	$f_1 = 1$		Klasse F

### Dose conversion coefficient and radiotoxicity equivalent for workers (w) and members of the public (b)

	Ingestie	Inhalatie	
	$f_1 = 1$	F	
$e(50)(w)$	$6,2 \times 10^{-9}$	$3,0 \times 10^{-9}$	Sv/Bq
$A_{Re}(w)$	$1,6 \times 10^8$	$3,3 \times 10^8$	Bq
$e(50)(b)$	$6,2 \times 10^{-9}$	$2,1 \times 10^{-9}$	Sv/Bq
$A_{Re}(b)$	$1,6 \times 10^8$	$4,8 \times 10^8$	Bq

### Data for total body counting (after single intake)

Time (d)	Total body activity (Bq per Bq intake)	
0,25	$9,9 \times 10^{-1}$	$7,4 \times 10^{-1}$
1	$9,8 \times 10^{-1}$	$6,0 \times 10^{-1}$
2	$9,6 \times 10^{-1}$	$5,1 \times 10^{-1}$
3	$9,3 \times 10^{-1}$	$4,7 \times 10^{-1}$
5	$8,9 \times 10^{-1}$	$4,4 \times 10^{-1}$
7	$8,5 \times 10^{-1}$	$4,2 \times 10^{-1}$

**Handboek Radionucliden [Radionuclides Handbook], A.S. Keverling Buisman (2nd edition, 2007), pp. 164-165,  $^{131}\text{I}$  data**

# $^{131}\text{I}$

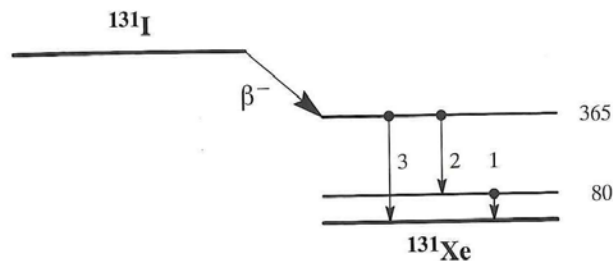
# $Z = 53$

### Halveringstijd en vervalconstante

$$T_{1/2} = 8,021 \text{ d} = 6,93 \times 10^5 \text{ s}$$

$$\lambda = 1,00 \times 10^{-6} \text{ s}^{-1}$$

### Vervalschema (vereenvoudigd)



### Belangrijkste uitgezonden straling

Straling	$y \text{ (Bq}\cdot\text{s)}^{-1}$	$E \text{ (keV)}$
$\beta^-$	0,894	192   606
$\gamma_1$	0,026	80
ce K $\gamma_1$	0,036	46
$\gamma_2$	0,061	284
$\gamma_3$	0,812	365

### Bronconstanten

Kermatempo in lucht	$k = 0,052 \text{ } \mu\text{Gy/h per MBq/m}^2$
Omgevingsdosisequivalenttempo	$h = 0,066 \text{ } \mu\text{Sv/h per MBq/m}^2$

### Diversen

Specifieke activiteit	$A_{\text{sp}} = 4,60 \times 10^{15} \text{ Bq/g}$
Vrijstellingsgrenzen	$C_v = 10^2 \text{ Bq/g}$ en $A_v = 10^6 \text{ Bq}$
Huidbesmetting	$H_{\text{huid}} = 4 \times 10^{-10} \text{ Sv/s per Bq/cm}^2$
Wondbesmetting; Injectie	$e(50) = 2,2 \times 10^{-8} \text{ Sv/Bq}$
Vervoer	$A_1 = 3 \text{ TBq}$
	$A_2 = 0,7 \text{ TBq}$

### Productie en toepassingen

Het radionuclide  $^{131}\text{I}$  is een belangrijk splijttingsproduct. Het wordt veelvuldig toegepast in de diagnostische en therapeutische nucleaire geneeskunde.

N = 78

131I

**Metabolic model**

Iodine in the blood: 70% direct excretion, 30% to the thyroid. Thyroidal iodine has a half-life of 80 days from the thyroid it spreads to other organs. Half-life in the other organs is 12 days. 1/10 of the organic iodine is excreted immediately with the faeces, the other 90% returns to the transfer compartment. So eventually the biological half-life in the thyroid is 90 days.

**Ingestion and lung clearance classes**

## Ingestie

Alle verbindingen  $f_1 = 1$ 

## Inhalatie

Damp (I <sub>2</sub> )	$f_1 = 1$	Klasse SR-1
Damp (CH <sub>3</sub> I)	$f_1 = 1$	Klasse SR-1 70% depositie
Overige verbindingen	$f_1 = 1$	Klasse F

**Dose conversion coefficient and radiotoxicity equivalent for workers (w) and members of the public (b)**

	Ingestie $f_1 = 1$	Inhalatie F	Inhalatie I <sub>2</sub>	Inhalatie CH <sub>3</sub> I	
$e(50)(w)$	$2,2 \times 10^{-8}$	$1,1 \times 10^{-8}$	$2,0 \times 10^{-8}$	$1,5 \times 10^{-8}$	Sv/Bq
$A_{Re}(w)$	$4,5 \times 10^7$	$9,1 \times 10^7$	$5,0 \times 10^7$	$6,7 \times 10^7$	Bq
$e(50)(b)$	$2,2 \times 10^{-8}$	$7,6 \times 10^{-9}$	$2,0 \times 10^{-8}$	$1,5 \times 10^{-8}$	Sv/Bq
$A_{Re}(b)$	$4,5 \times 10^7$	$1,3 \times 10^8$	$5,0 \times 10^7$	$6,7 \times 10^7$	Bq

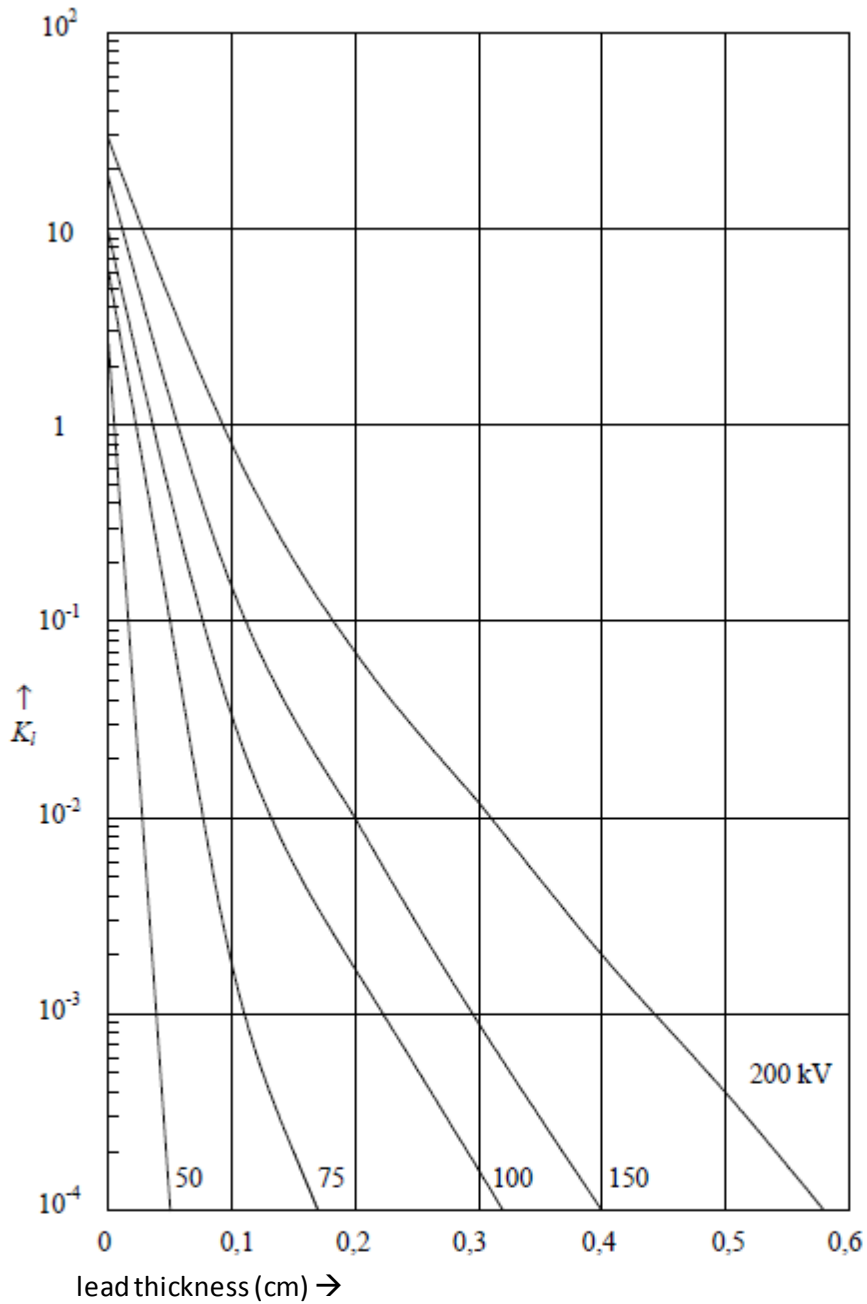
**Data for thyroid count (after single intake)**

Time (d)	Activity in thyroid (Bq per Bq intake)			
	$f_1 = 1$	F	I <sub>2</sub>	CH <sub>3</sub> I
0,25	$6,0 \times 10^{-2}$	$5,2 \times 10^{-2}$	$1,1 \times 10^{-1}$	$1,0 \times 10^{-1}$
1	$2,4 \times 10^{-1}$	$1,2 \times 10^{-1}$	$2,3 \times 10^{-1}$	$1,8 \times 10^{-1}$
2	$2,5 \times 10^{-1}$	$1,2 \times 10^{-1}$	$2,2 \times 10^{-1}$	$1,7 \times 10^{-1}$
3	$2,3 \times 10^{-1}$	$1,1 \times 10^{-1}$	$2,0 \times 10^{-1}$	$1,6 \times 10^{-1}$
5	$1,9 \times 10^{-1}$	$9,0 \times 10^{-2}$	$1,7 \times 10^{-1}$	$1,3 \times 10^{-1}$
7	$1,6 \times 10^{-1}$	$7,5 \times 10^{-2}$	$1,4 \times 10^{-1}$	$1,1 \times 10^{-1}$

**Table 3: Tissue weighting factors (derived from ICRP-60)**

<b>Tissue or Organ</b>	<b>Tissue weighting factor (<math>W_T</math>)</b>
Gonads	0.20
Red bone marrow	0.12
Large intestine	0.12
Lungs	0.12
Stomach	0.12
Breast tissue	0.05
Bladder	0.05
Liver	0.05
Oesophagus	0.05
Thyroid gland	0.05
Skin	0.01
Bone surface	0.01
Other organs	0.05

**Figure 3: Kerma rate free in air of X-radiation through lead**



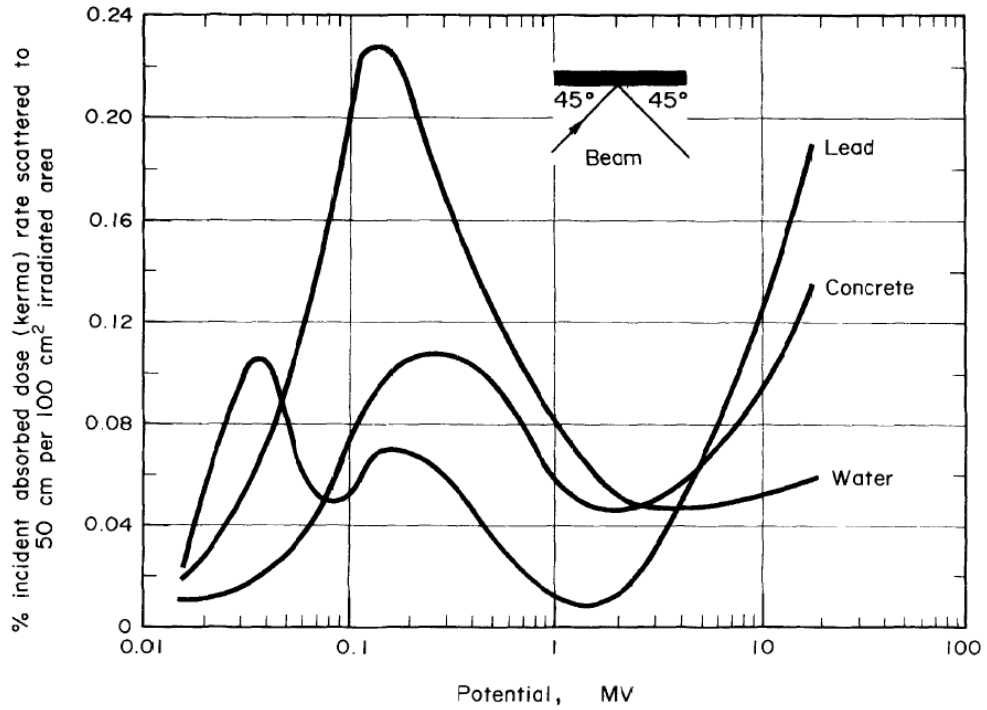
Yield of a broad beam of X-rays through lead, expressed in kerma  $K_1$  free in air: in  $\text{mGy} \cdot \text{m}^2 \cdot \text{mA}^{-1} \cdot \text{min}^{-1}$  as a function of high voltage in kV.

Constant high voltage; tungsten target; 2 mm A1 pre-filtering

Intersection with the Y axis, so kerma without shielding ( $\text{mGy} \cdot \text{m}^2 \cdot \text{mA}^{-1} \cdot \text{min}^{-1}$ ):

50 kV 2.6	75 kV 6.1	100 kV 9.6	150 kV 18.3	200 kV 28.7
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**Figure 4: Scattering fraction of kerma in air**



Variation with potential of the absorbed dose (kerma) rate measured in air due to X-rays scattered at 90° from various materials (ICRP 33, Figure 21)